

**I. Multiple Choice:** circle the one best answer; 2 pts. each.

1. A "fertile" soil is defined (specifically) as a soil with--
  - A. no limiting factors for plant growth.
  - B. large numbers of soil organisms such as worms, insects, etc.
  - C. a relative abundance of available nutrients and no excessive acids, salts, etc.
  - D. no need for any additional fertilizer or other amendment to give maximum yields
  - E. high concentrations of humus and other essential growth factors
2. A major advantage of using organic fertilizers, compared to commercial synthetics, is--
  - A. they have a high analysis compared to other fertilizers
  - B. their nutrients are highly soluble and readily plant-available
  - C. they add C to the soil, increasing humus levels, and also add other nutrients
  - D. only small amounts need to be applied, which can lower fertilizer costs
  - E. none of the above are true
3. Gypsum ( $\text{CaSO}_4$ ) is--
  - A. less soluble than lime.
  - B. produced by removing  $\text{SO}_2$  (acid rain) from power plants.
  - C. useful for raising the pH of soils.
  - D. often causes a large decrease in soil pH.
  - E. too expensive for common agricultural use.
4. When a low (20:1) C:N residue is mixed into a soil, which of the following would occur?
  - A. N is immobilized rapidly, but eventually released as the C:N ratio is lowered over time.
  - B. N is readily mineralized due to the rapid decomposition by bacteria.
  - C. some N is mineralized into solution, but is quickly nitrified and made unavailable.
  - D. the C:N ratio increases rapidly as  $\text{NH}_4$  and  $\text{NO}_3$  are removed from the soil.
  - E. very little decomposition occurs because this residue is probably mostly lignin.
5. A major role that soil organisms play in cycling of many nutrients is--
  - A. enhancing the loss of nutrients by encouraging leaching.
  - B. removing nutrients from cycles by permanently immobilizing them.
  - C. adding large amounts of new nutrients to cycles by weathering soil minerals.
  - D. converting organic nutrients back to inorganic forms during decomposition.
  - E. more than one of the above.
6. Which of the following is a result of liming an acid soil?
  - A. aluminum solubility increases
  - B.  $(\text{H}^+)$  concentration in solution decreases
  - C. % base saturation decreases
  - D. soil pH decreases
  - E. cation exchange capacity decreases
7. Which one of the general forms of nutrients in soils is the *least* plant-available?
  - A. ions in the soil solution
  - B. strongly adsorbed nutrients
  - C. nutrients held within mineral structures
  - D. ions held on CEC and AEC sites
  - E. all of the above are equally plant-available

8. *Permanent* charge on clay minerals is caused by--
- replacement of one cation by another of lower charge during mineral formation
  - cations being strongly held on the mineral surface (adsorption)
  - adsorption of hydroxyl (-OH) groups from solution under alkaline conditions
  - protonation of functional groups in the interlayer space
  - changes in the structure of clays due to weathering over geologic time
9. The best way to sample soil from a field to evaluate fertilizer needs is to--
- lay out a grid on the field and sample at evenly spaced intervals.
  - take a shovelful of soil from near the center of the field.
  - take a composite sample of 10-15 random locations within the field..
  - sample along a transect from one end of the field to the other.
  - any of the above would be OK.

**II. Matching:** write the letter of the best answer in the blank; use each answer only once. 1 pt.

10. Clay Minerals:
- |   |                          |
|---|--------------------------|
| _____ low CEC, 1:1, common in highly weathered soils                | A. illite (hydrous mica) |
| _____ 2:1, high octahedral charge; large shrink/swell capacity      | B. vermiculite           |
| _____ 2:1, v. high tetrahedral charge; limited shrink/swell         | C. montmorillonite       |
| _____ initial weathering product of muscovite; partial K interlayer | D. kaolinite             |
|   | E. chlorite              |
11. N cycle:
- |                                      |                         |
|--------------------------------------|-------------------------|
| _____ $N_2(g) \rightarrow$ organic N | A. immobilization       |
| _____ $NH_4^+ \rightarrow$ organic N | B. denitrification      |
| _____ organic N $\rightarrow NH_4^+$ | C. symbiotic N fixation |
| _____ $NO_3^- \rightarrow N_2(g)$    | D. nitrification        |
|                                      | E. mineralization       |

12. Laws/Principles:

- |  |                        |
|--|------------------------|
| _____ states how different growth factors affect plant yield     | A. Stokes'             |
| _____ explains how levels of inputs and outputs determine profit | B. Liebig's            |
| _____ describes rate of settling of particles in water           | C. Le Chatelier's      |
| _____ describes how chemical equilibria respond to disturbance   | D. Diminishing Returns |

**13. Fill-In:** write a single word or phrase that matches the definition; 1 pt.

- \_\_\_\_\_ pH of a solution containing  $10^{-4}$  moles/L HCl
- \_\_\_\_\_ symbiotic fungi that enhance plant root uptake of nutrients like P
- \_\_\_\_\_  $6 \times 10^{23}$  of something (like, atoms)
- \_\_\_\_\_ a pollutant element that behaves in soil similarly to micronutrient anions
- \_\_\_\_\_ element that can substitute for Si in tetrahedral structure of clay minerals
- \_\_\_\_\_ lab instrument used to measure cation (Ca, Mg) concentrations in solution
- \_\_\_\_\_ term for limestone that includes a significant amount of Mg
- \_\_\_\_\_ method used to measure acid concentrations in solutions
- \_\_\_\_\_ name or formula of most important charge-generating group on humus
- \_\_\_\_\_ genus of bacteria that can symbiotically convert  $N_2(g)$  into organic N
- \_\_\_\_\_ trade term referring the potassium, or fertilizer containing K

**III. Short Answer/Calculation:** write a short response to the following; do not restate the question, but give examples when necessary. For calculation, show all your work clearly.

**NOTE:** Ca: 40    Mg: 24    K:39    P: 31    O: 16    H: 1    N: 14    C: 12

14. List and briefly explain three ways that increasing levels of humus in soils tend to increase overall soil productivity. (3 pts.)
15. Show/explain how charge is developed on humus, and how pH affects this charge. (3 pts)
16. A soil contains the following exchangeable cations (in meq/100 g): Mg: 0.5; Ca: 2.3; K: 0.3; H: 0.5; Al: 1.9. Calculate:
- A. CEC (2 pts)
- B. %BS: (2 pts)
- C. lbs K/afs: (3 pts)
17. A plastic-coated clod weighing 44 g was immersed in water and found to displace 35 g. of water. (4 pts.)
- A. Calculate bulk density, showing *units*.
- B. Calculate % pore space
- C. If this sample came from a loamy A horizon, is this BD favorable value for plant growth?

18. A mechanical analysis on 50 g soil gave a 40-sec hydrometer reading of 18 g/L, and a 6-hr (second) reading of 8 g/L. Calculate the % sand, silt and clay in this sample, and estimate the textural class of the soil. (4 pts.)

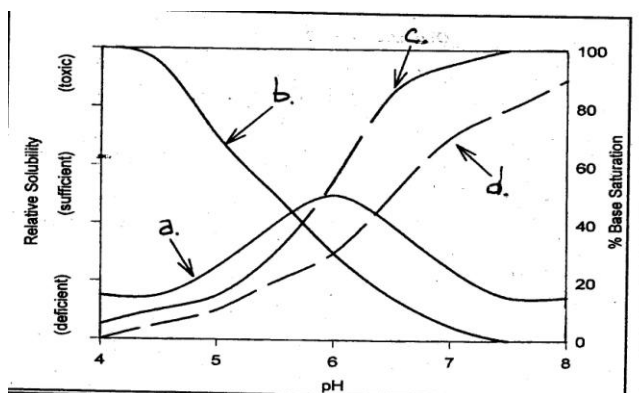
19. Convert  $1.55 \text{ g/cm}^3$  to  $\text{lbs/ft}^3$  [ $454\text{g}=1\text{lb}$ ,  $2.54 \text{ cm}=1 \text{ inch}$ ]. (3 pts)

20. In measuring CEC, cations are extracted from 2.5 g soil using 40 mL of salt solution. This solution is found to contain 1.7 meq/L of Ca. Calculate exchangeable Ca in meq/100 g. (3 pts)

21. For the graph at right, identify the nutrient or nutrient *group* that corresponds with each line:

- a. \_\_\_\_\_
- b. \_\_\_\_\_
- c. \_\_\_\_\_
- d. \_\_\_\_\_

(4 pts)



BONUS:

A. Nitrogen is NOT shown on the graph in 21. above. Explain briefly why not. (2 pts)

B. How many mL of 1 N acid would it take to completely neutralize 1 g of pure  $\text{CaCO}_3$ ? (2 pts)

C. What is your favorite micro-nutrient cation? \_\_\_\_ (1 pt)